

## SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR

## (AUTONOMOUS)

## B.Tech I Year I Semester Regular Examinations July-2021

**BASIC THERMOYNAMICS** 

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(A or	aculturs	al En	one	ering
LISI	rounding		Sinc	or mg

Max Marke 60

	Tin	ne: 3 hours	Max. M	larks: 6
		(Answer all Five Units $5 \times 12 = 60$ Marks)		
		UNIT-I		
1	9	What is quasi static process? What are its characteristics features?	Τ1	6M
T	a h	Explain about Thermodynamic Equilibrium		6M
	N	OR		UIVI
2	<b>a</b> Explain thermodynamics system surrounding and universal Distinguish between			
-		closed open isolated Systems		OIVI
	b	State the thermodynamic system control volume	L1	<b>4</b> M
		UNIT-II		
3	9	What are the Limitations of First laws of thermodynamics?	Ι1	6M
5	а 1	what are the Limitations of First laws of thermodynamics?		OIVI
	b	A system changes from state 1 to state 2 along the path 1a2 absorbs 75JK of hea	t L3	6M
		and does 30 KJ of work. The system is returned from state 2 to state 1 along the path 2b1 by doing a work of 10 KL Find and the best transformation of a state 1 along the path	1	
		201 by doing a work of 10 KJ. Find out the heat transfer along the path 201.		
1	0	Explain reversible and irreversible process	1.2	(M
7	a h	What are the different modes in which energy is stored in a system?	L2 I1	6M
	IJ	what are the different modes in which energy is stored in a system?		OIVI
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5	a	What is Avogadro's law?		6M
	b	State Internal Energy and Enthalpy of Gas.	Ll	6M
		OR		
6	a	Derive an expression for heat transfer during polytrophic process	L4	6M
	b	Air in a closed stationary system expands in a reversible adiabatic process from 0.4	> L3	6M
		MPa, 15°C to 0.2 MPa. Find the final temperature, and per kg of air, the heat	t	
		transferred, and the work done.		
_				
7	a	Derive an expression for the thermal efficiency of Sterling cycle and draw P-V & T	- L4	6M
	1	S diagrams.	т.а	
	D	Find the change in enthalpy steam, initial pressure 15 bar and 0.95 then it will reach	L3	6 M
		25 bar and 400 temperature. By using momer diagram.		
8	9	Derive an expression for the thermal efficiency and mean effective pressure of a	<b>. I</b> 4	6M
0	а	Otto cycle by drawing PV and TS diagrams	1 1/4	UIVI
	b	Find the change in enthalpy steam initial pressure 12 har and 200°C then it wil	1 13	6M
	N	reach 0.95 in isentronic process	1 115	UNI
		INIT V		
0		State the methods of increasing the thermal afficiency of Danking avala	Т 1	(M
9	a h	Derive the expression for efficiency of Derkine cycle with D.V. T.S. Discreme		OIVI (M
	D	Derive the expression for efficiency of Rankine cycle with P-V, 1-S Diagrams.	L4	0111
10	9	State the advantages and disadvantages of a Debast avala	T 1	6M
U	a h	A Steam nower plant operates at a pressure of 15 har 2000C expands through a high		6M
	IJ	pressure turbine. It is reheated at a pressure of 4 bars to 3000 C and expands through		UIVI
		the low pressure turbine to a condenser pressure of 0.1 har. Determine work dow	2	
		and cycle efficiency.	-	
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